Flue Gas Purification Utilizing SOx/NOx Reactions During Compression of CO₂ Derived from Oxyfuel Combustion (Oxy – T – Fired)

(NETL Cooperative Agreement No. DE-NT0005309)

Department of Energy/National Energy Technology Laboratory (DOE/NETL) 2010 CO2 Capture Technology R&D Meeting, 13-17 September 2010 Pittsburgh, PA, USA

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Air Products and Chemicals, Inc.

Who Is Air Products?

- Global atmospheric, process and specialty gases, performance materials, equipment and services provider
 - Serving industrial, energy, technology and healthcare markets worldwide
- Fortune 500 company
- Known for our innovative culture and operational excellence
- Safety leader in the chemical industry
- Capture techniques
 - Based upon wide experience in ASU, HyCO, combustion applications, cryogenic separations, compression & CO₂ handling
 - Promising proprietary developments point to reductions in cost of CO₂ capture





Agreement Period of Performance & Cost Share

- Period of Performance:
 - 1 October 2008 30 September 2010

- Air Products
- NETL Cost Share:
- Overall Project Total:

- \$ 251,000 (20%)
- \$ 1,003,995 (80%)
- \$ 1,254,995

- Project Participants: Air Products
- Host Site: Alstom Power Power Plant Laboratories
 Boiler Simulation Facility in Windsor, CT.



Technology Fundamentals

- What is the technology?
- Current status of technology
- Design for the PDU (process development unit)
- Results from PDU campaigns
- Next Steps



Oxyfuel CO₂ Purification

- Oxyfuel combustion of coal produces a flue gas containing:
 - CO₂ + H₂O
 - Any inerts from air in leakage or oxygen impurities
 - Oxidation products and impurities from the fuel (SO_x , NO_x , HCl, Hg, etc.)
- Purification requires:
 - Cooling to remove water
 - Compression to 30 bar
 - Integrated SOx/NOx/Hg removal
 - Low Temperature Purification
 - Low purity, bulk inerts removal
 - High purity, Oxygen removal
 - Compression to pipeline pressure



NOx SO₂ Reactions in the CO₂ Compression System

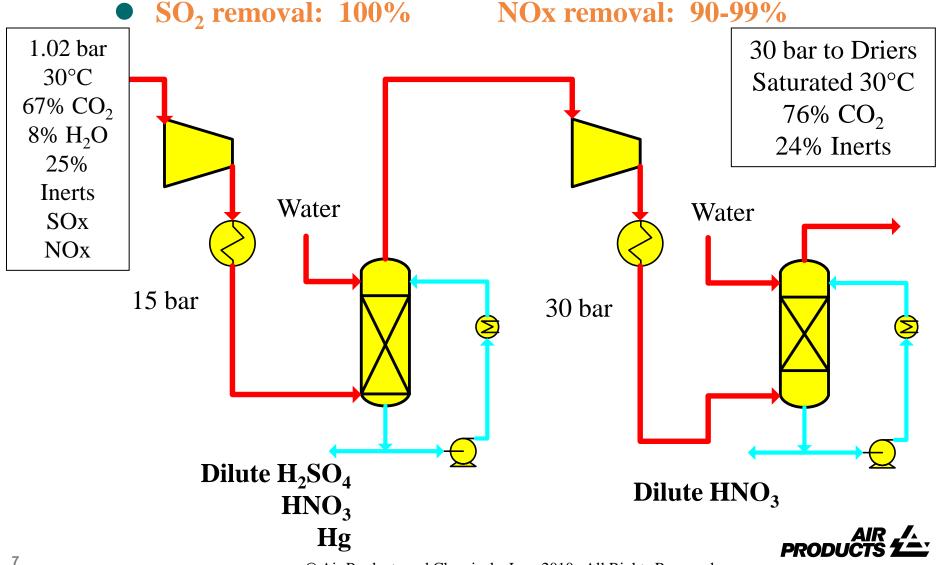
- We realized that SO₂, NOx and Hg can be removed in the CO₂ compression process, in the presence of water and oxygen.
- SO₂ is converted to Sulfuric Acid, NO₂ converted to Nitric Acid:

$-$ NO + $\frac{1}{2}$ O ₂	=	NO_2	(1) Slow
- 2 NO ₂	=	$N_2\bar{O_4}$	(2) Fast
$- 2 NO_{2}^{-} + H_{2}O$	=	$H\overline{NO}_2 + HNO_3$	(3) Slow
-3 HNO2	=	$HNO_{3} + 2 NO + H_{2}O$	(4) Fast
$- NO_2 + \overline{SO}_2$	=	$NO + SO_3$	(5) Fast
$- SO_3 + H_2O$	=	H ₂ SO ₄	(6) Fast

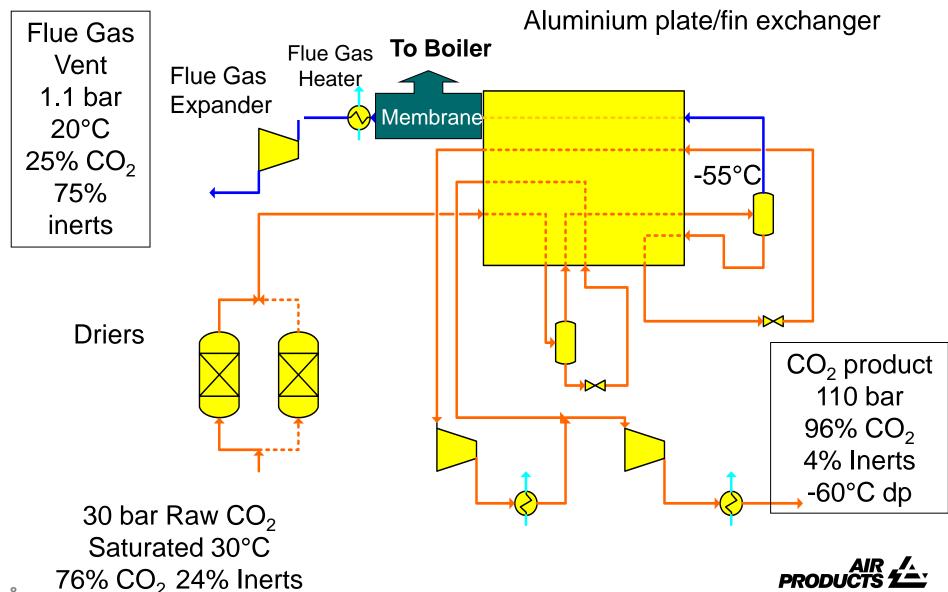
- Rate increases with Pressure to the 3rd power
 - only feasible at elevated pressure
- Little Nitric Acid is formed until all the SO₂ is converted
- Pressure, reactor design and residence times, are important.



Air Products' CO₂ Compression and Purification System: Removal of SO₂, NO_x and Hg



Air Products' System: Inerts removal and compression to 110 bar



SOx/NOx Removal – Key Features

- Adiabatic compression to 15 bar:
 - No interstage water removal
 - All Water and SOx removed at one place
- NO acts as a catalyst
 - NO is oxidized to NO₂ and then NO₂ oxidizes SO₂ to SO₃: The Lead Chamber Process
- Hg will also be removed, reacting with the nitric acid that is formed
- FGD and DeNOx systems are not required for emissions or CO₂ purity
 - SOx/NOx removed in compression system
 - Low NOx burners are not required for oxyfuel combustion



Path to from Lab

to Demo



 $160 \, kW_{th}$ oxy-coal rig



 $15\,MW_{th}$ oxy-coal combustion unit

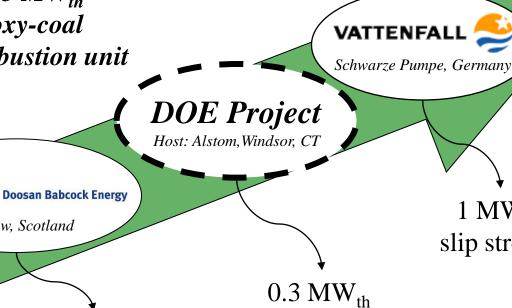
Renfrew, Scotland



30 MW_{th} oxy-coal pilot plant



50-250 MW_e oxy-coal **Demonstration**



 $0.3~\mathrm{MW_{th}}$ slip stream



 1 MW_{th}

slip stream



Imperial College London London

Batch

DOOSAN

 6 kW_{th} slip stream

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Project objectives

 To purify the CO₂ derived from oxy-coal combustion by utilizing the SOx / NOx reactions that will occur during CO₂ compression

Phase 1

Design and Construction of Reactor System for Purification of CO₂ from Oxy-Coal Combustion

- The Phase I objectives include the design, construction, and commissioning of a 15 bar reactor system for removal of SOx /NOx from actual oxy-coal derived, CO₂-rich flue gas.
- The system will be designed to cool an oxy-coal combustion flue gas slip stream(~0.35 MW_{th} flow rate equivalent), compress from 1 bar to 15 bar and react within a 15 bar column the SOx/NOx present in the CO₂ rich flue gas.



Project objectives (continued)

Phase II

Evaluate Robustness of Reactor Performance for Purification CO₂ from Oxy-Coal Combustion

- The Phase II objectives include further evaluations of the reaction process using oxy-coal derived flue gas generated by the host site (Alstom).
 - Evaluate the performance of the reactor based on the reactor effluents for different reactor pressures as well as water recycle rates
 - Characterize the reactor effluents (both liquid and gaseous) to assess any change in reactor performance
- Air Products will develop an engineering model to describe the
 15 bar purification reactor performance.
 - Perform a sensitivity analysis using said model to elucidate those parameters most critical to performance

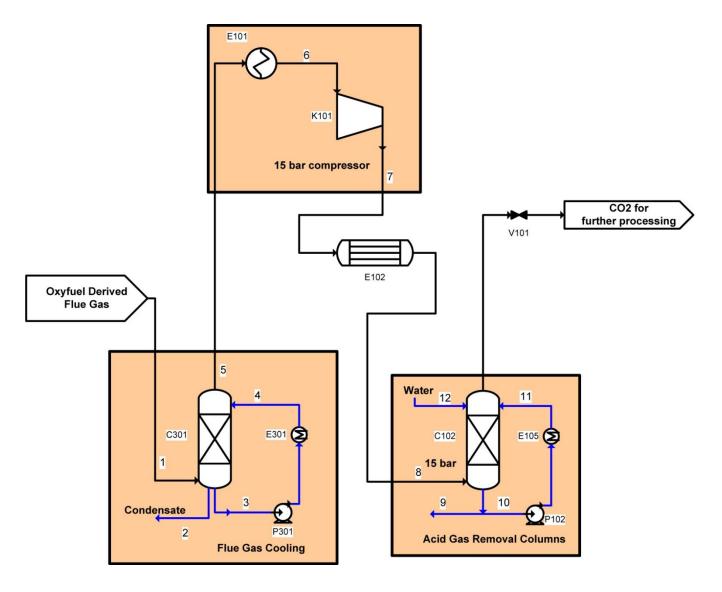


Milestones / Schedule

- Initiate Construction of Reactor System
 - Planned Date: Complete
- Initiate Testing of Reactor System
 - Planned Date: Complete
- Evaluate Performance of Reactor Based Flue Gas
 - Planned Date: Complete
- Develop Engineering Model and Perform Sensitivity Analysis
 - Planned Date: September 30, 2010



Current Process Flow Diagram



PDU (process development unit)



DOE Project: Air Products' Sour Compression PDU



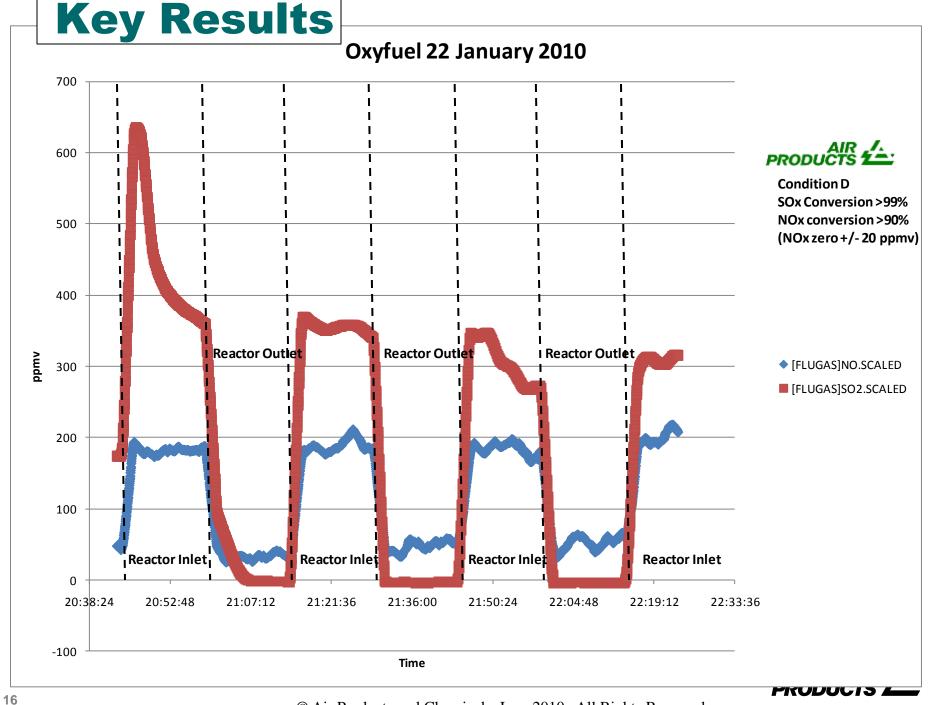


Side View of PDU

- 1st campaign Jan 2010
- 2nd campaign April-May 2010

Acid Reactor (C102)



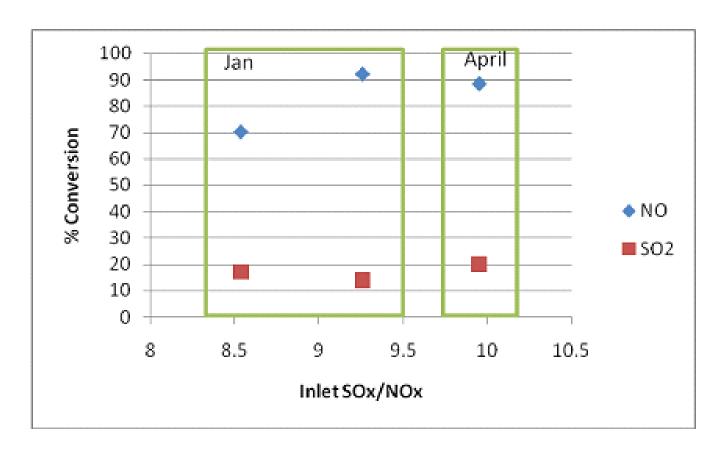


Results

- For the overall process, total SO₂ removal was 20-100 % (based on gas compositions).
- For the overall process, total NOx removal was 60-90 % (based on gas compositions).
- The effects of variations in the SO₂/NOx feed ratio, column pressure, gas flowrate and liquid recirculation on the reactor performance were elucidated. Process performance was most sensitive to SO₂/NOx feed ratio, over the range of parameter values investigated.
- SO₂ was removed from the flue gas through both sulfite and sulfate mechanisms.



Comparison of NO and SO₂ conversions



- Reproducing test conditions shows consistent results
- Confirms process and measurement reproducibility



Impact of column operating parameters on SOx and NOx conversion

↑ in Operating Parameter	SO _x Conversion	NO _x Conversion
Column pressure	↑	↑
Column gas flow-rate	\	\
Column recirculation liquid flow-rate	1	↑
Column fresh make-up water flow-rate	1	↑
Column inlet SO _x /NO _x ratio	\	<u> </u>



Advantages

- FGD and DeNOx systems are not required for emissions or CO₂ purity
 - SOx/NOx removed in compression system
 - Low NOx burners are not required for oxyfuel combustion
- Oxygen can be removed to produce EOR-grade CO₂
- No penalty if CO₂ is required as a liquid
- Vent stream is clean, at pressure and rich in CO₂ (~25%) and O₂ (~20%)
 - Polymeric membrane unit selective for CO₂ and O₂ in vent stream will recycle CO₂ and O₂ rich permeate stream to boiler.
 - CO₂ Capture increase to >97%
 - ASU size/power reduced ~5%



Challenges

- Optimization of SOx, NOx, & Hg removal
- Reaction kinetics / equilibrium
- Fouling / impurities effects
- Materials of construction
- Byproduct streams H₂SO₄,
 HNO₃, Hg species,...
- Burners must be demonstrated with flue gas recycle
- Minimization of parasitic power for O₂ supply and CO₂ compression / purification

PDUCPU Pilot Plant

Boiler OEMs

Reference Plants
Design
FEED Studies



Path to from Lab

to Demo



 $160 \, kW_{th}$ oxy-coal rig

DOOSAN

Batch



 $15\,MW_{th}$ oxy-coal combustion unit



30 MW_{th} oxy-coal pilot plant

 $0.3~\mathrm{MW_{th}}$

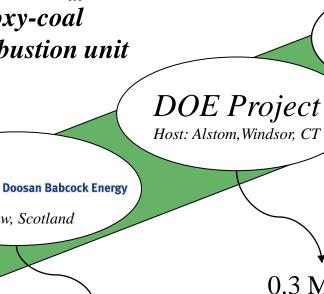
slip stream

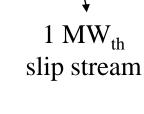


50-250 MW_e oxy-coal **Demonstration**

VATTENFALL

Schwarze Pumpe, Germany







Imperial College

 6 kW_{th} slip stream

Renfrew, Scotland



Cylinder fed

bench rig

Next Steps

- Complete final report for PDU / small scale testing of SOx/NOx removal
- Move to pilot scale for CO₂ purification and compression



- Scale up to Pilot: Underway
- Demonstration on stream: 2015
- Commercialization: 2017-2020



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